

Synthesis and Evaluation of a Nanoglobular Dendrimer 5-Aminosalicylic Acid Conjugate with a Hydrolyzable Schiff Base Spacer for **Treating Retinal Degeneration**

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ABSTRACT



Biocompatible dendrimers with well-defined nanosizes are increasingly being used as carriers for drug delivery. 5-Aminosalicylic acid (5-ASA) is an FDA-approved therapeutic agent recently found effective in treating retinal degeneration of animal models. Here, a water-soluble dendrimer conjugate of 5-ASA (AGFB-ASA) was designed to treat such retinal degeneration. The drug was conjugated to a generation 2 (G2) lysine dendrimer with a silsesquioxane core (nanoglobule) by using a hydrolyzable Schiff base spacer. Incubation of nanoglobular G2 dendrimer conjugates containing a 4-formylbenzoate (FB) Schiff base spacer in pH 7.4 phosphate buffers at 37 °C gradually released 5-ASA. Drug release from the dendrimer conjugate was significantly slower than from the low molecular weight free Schiff base of 5-ASA (FB-ASA). 5-ASA release from the dendrimer conjugate was dependent on steric hindrance around the spacer. After intraperitoneal injection, the nanoglobular 5-ASA conjugate provided more effective 7-day protection against light-induced retinal degeneration at a reduced dose than free 5-ASA in Abca4^{-/-}Rdh8^{-/-} mice. The dendrimer 5-ASA conjugate with a degradable spacer could be a good candidate for controlled delivery of 5-ASA to the eye for treatment of retinal degeneration.

KEYWORDS: dendrimer · 5-aminosalicylic acid · Schiff base · drug delivery · controlled release

t is well known that the aldehyde alltrans-retinal (atRAL) is a major intermediate in the visual cycle.^{1–5} Photoactivated rhodopsin releases atRAL, which is subsequently transported by an ATP-binding cassette transporter 4 (Abca4) and reduced to all-trans-retinol by all-trans-retinol dehydrogenases located in photoreceptor cells. Continuous regeneration of the 11-cis chromophore from atRAL is essential for both renewal of light-sensitive visual pigments required for vision and photoreceptor survival in the vertebrate retina.⁵

Disruptions in the conversion or clearance of atRAL in photoreceptors can cause accumulation of this reactive atRAL aldehyde and its toxic condensation products with eventual manifestations of retinal dystrophy, including human retinal degenerative diseases such as Stargardt's disease and age-related macular degeneration. Thus, it appears that accumulation of atRAL is one of the key factors initiating retinal photodamage characterized by progressive retinal cell death evoked by both acute and chronic light exposure.

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One of the pharmacological innovations to protect against photodamage mediated by atRAL is the use of aldehyde-reactive amines to reversibly sequester atRAL as a Schiff base, thereby lowering its tissue concentration.⁴ Slow release of atRAL from the Schiff base allows the retinoid to flow back into the retinoid cycle without affecting visual chromophore regeneration and phototransduction.^{1,5} As an FDA-approved compound, 5-aminosalicylic acid (5-ASA) containing a primary amine group has a high potential for preventing light-induced retinopathy in a mouse model of human retinal diseases, the Abca4/Rdh8 (Abca4-/-Rdh8-/-) double knockout mouse.⁵ However, free 5-ASA exhibits rapid pharmacokinetics $(t_{1/2} = 0.4 \text{ to } 2.4 \text{ h})^6$ with poor tissue delivery and therapeutic efficacy due to its low water solubility (<0.9 mg/mL). The use of a suitable drug delivery system can provide better and prolonged therapeutic efficacy at reduced doses. Thus we aimed to design and develop an effective delivery system for sustained delivery of 5-ASA to the retina.

Intraocular drug delivery has been a major challenge due to the unique anatomy and physiology of the eye.⁷⁻¹⁰ Due to blood/ocular barriers, conventional drug delivery systems, including solutions, suspensions, and gels, can deliver only a small percentage of administrated drugs across the cornea, conjunctiva, and sclera to reach affected sites of the retina, especially the posterior segments. As a result, relatively large doses of drugs must be frequently administrated to achieve and maintain therapeutic concentrations in the eyes. Rapid development of biomaterials and increasing understanding of ocular drug absorption and disposition mechanisms have created new possibilities for ocular drug delivery.^{11,12} Various new drug delivery systems, including dendrimers, microemulsions, mucoadhesive polymers, hydrogels, iontophoresis, microneedles, and prodrugs, have been employed to improve the pharmacokinetics and pharmacodynamics of ocular therapeutics.^{13–20} These novel delivery systems offer many advantages over conventional formulations due to their improved drug delivery profiles and reduced drug toxicity. Among such advanced delivery systems, polymer-based nanocarriers appear highly attractive and are being extensively investigated.^{21,22} Compared with linear polymers, dendrimers are well-defined, globular highly branched macromolecules with a large number of functional groups at their surface. Moreover, their size, molecular weight, and surface properties can be easily controlled. Dendrimers have attracted much attention in drug delivery due to these unique structural properties and good water solubility.^{23–30} Thus a relatively large number of drug molecules can be conjugated to dendrimers with proper chemical spacers to achieve controlled and sustained drug release.

Here we designed and synthesized a G_2 nanoglobular conjugate of 5-ASA with a degradable spacer.

Nanoglobules are a new class of lysine dendrimers with a silsesquioxane core.^{31,32} The G₂ nanoglobule has 32 surface amino groups for drug conjugation. A Schiff base spacer is designed to release the drug *via* hydrolysis under physiological conditions.^{33,34} Drug release from the conjugate was investigated *in vitro* and was compared with that of the free Schiff base of 5-ASA. Finally, the therapeutic efficacy of the conjugate was preliminarily tested in the *Abca4^{-/-}Rdh8^{-/-}* mouse⁴ model *in vivo*.

RESULTS

Synthesis of the Dendrimer Conjugate AGFB-ASA. The G₂ lysine dendrimer with an octa(3-aminopropyl)silsesquioxane (nanoglobule) was synthesized through a divergent approach by reacting octa(3-aminopropyl)silsesquioxane or a lower generation dendrimer with (Boc)₂-L-lysine-OH pentafluorophenol active ester in the presence of diisopropylethylamine. The G₂ nanoglobule was obtained in 98.0% yield and high purity. The structure of the G₂ nanoglobular dendrimer is shown in Figure 1a. Standard liquid phase synthesis was used to synthesize the G₂ nanoglobular 5-ASA Schiff base conjugate, AGFB-ASA, and the scheme of this synthesis is shown in Figure 1b. All intermediates and the final product were confirmed by MALDI-TOF mass spectrometry and ¹H NMR spectroscopy. The MALDI-TOF mass spectrum and proton NMR spectrum of the final product AGFB-ASA are shown in Figure 2b and c. The chemical shift at 0.63 was derived from the 8-methylene proton, and its integral value was assigned as 16. The integral value of the imine proton at a chemical shift of 8.63 was 5.6, whereas that of the benzene proton on the linker and 5-ASA at the chemical shift from 7.93 to 8.07 was 35.3. Integral values of specific group protons confirmed the assignment of a 1:5:5 composition of the dendrimer:spacer:5-ASA with approximately five drug molecules on each G₂ dendrimer molecule. The number of attached 5-ASA molecules was also determined by measuring the absorbance of the Schiff base (imine aromatic moiety) in DMSO by UV spectroscopy at 325 nm. The calculated number of 5-ASA residues in the dendrimer conjugate estimated by UV spectroscopy was 5.8, similar to the number (ca. 5) determined by proton NMR spectroscopy. The dendrimer drug conjugate had good water solubility. The free Schiff base conjugate of 5-ASA with 4-formylbenzoic acid was also synthesized and used as a control, Figure 1d.

Drug Release Studies. The release of 5-ASA from the dendrimer conjugate AGFB-ASA was evaluated in phosphate buffers at pH 7.4 and 37 °C along with FB-ASA as a control. Figure 3 shows the time course of UV spectral change from the conjugates incubated in phosphate buffer. The maximum UV absorption of the Schiff base in PBS was at 325 nm at pH 7.4. Degradation of the Schiff base spacer was reflected by the decrease of absorption at 325 nm over time. Schiff base degradation of AGFB-ASA was much slower

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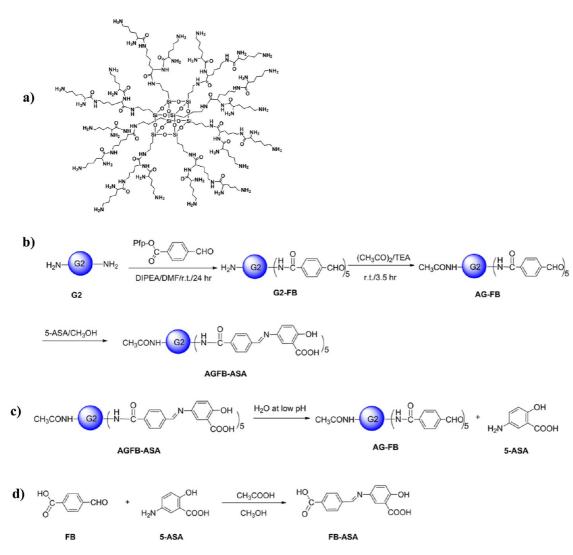


Figure 1. The AGFB-ASA conjugate: structure, synthesis, and release of 5-ASA. (a) Structure of the nanoglobular dendrimer G_2 ; (b) scheme of AGFB-ASA conjugate synthesis; (c) 5-ASA release from the conjugate; (d) scheme of FB-ASA conjugate synthesis.

than that of the free Schiff base conjugate FB-ASA at pH 7.4. Drug release from both conjugates was calculated from their UV absorption at 325 nm as depicted in Figure 4a. At physiological pH 7.4, the dendrimer conjugate AGFB-ASA also showed a more controllable release pattern compared to the free Schiff base FB-ASA with the identity of the released drug further confirmed by HPLC (Figure 4b). The HPLC elution times for 5-ASA released from AGFB-ASA and FB-ASA were 3.335 and 3.276 min, respectively, consistent with the retention time of the 5-ASA standard (3.331 min).

Effects of AGFB-ASA on Preventing Light-Induced Retinal Degeneration. The therapeutic efficacy of the AGFB-ASA conjugate was investigated in 4-week-old $Abca4^{-/}$ $^{-}Rdh8^{-/-}$ mice, after acute strong light exposure. In this animal model of Stargardt's disease, strong light exposure results in accumulation of atRAL which can cause retinal degeneration. The AGFB-ASA conjugate after intraperitoneal administration should release free 5-ASA and reduce the accumulation of atRAL in the retina,

preventing light-induced retinal degeneration. Schematic representation of the experimental design is presented in Figure 5a.

Electroretinograms (ERGs) were recorded to evaluate retinal functions after drug pretreatment with either free 5-ASA or conjugate AGFB-ASA at an equivalent dose of 0.5 mg of 5-ASA per mouse and subsequent light illumination of *Abca4^{-/-}Rdh8^{-/-}* mice at 4 weeks of age. ERG responses of conjugate AGFB-ASApretreated light-illuminated mice (AGFB-ASA-LI) were maintained, as the no light-illuminated mice (NLI), in contrast to 5-ASA-pretreated light-illuminated mice (free 5-ASA-LI), displayed reduced ERG peak amplitudes (p < 0.05, Student's t test) (Figure 5b).

Quantitative morphometry of ONL thickness measured by OCT imaging was examined in mice treated with 5-ASA and the conjugate at doses equivalent to 0.5 and 1.0 mg of 5-ASA per mouse. AGFB-ASA-pretreated light-illuminated $Abca4^{-/-}Rdh8^{-/-}$ mice evidenced a more preserved ONL thickness than free

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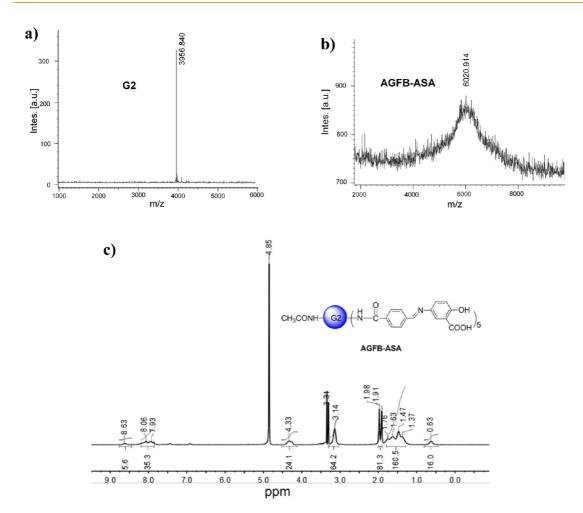


Figure 2. Characterization of the conjugate. (a) MALDI-TOF spectrum of G₂; (b) MALDI-TOF spectrum of the final conjugate, AGFB-ASA; (c) ¹H NMR spectrum of the AGFB-ASA conjugate.

5-ASA-pretreated light-illuminated mice, which failed to exhibit any protection against light-induced retinal degeneration at 0.5 mg of 5-ASA per mouse (Figure 5c). Of note, drug treatments with the 1.0 mg dose in both formulations prevented retinal degeneration with greater protection conferred by the AGFB-ASA (Figure 5d). Representative OCT images from light-illuminated 4-weekold mice pretreated with conjugate free 5-ASA or AGFB-ASA are shown in Figure 5e. AGFB-ASA provided more effective protection against light-induced retinal degeneration at the reduced dose than the free drug.

DISCUSSION

The G₂ nanoglobule was synthesized by a modified synthetic method that employed a pentafluorophenolactivated ester of Boc-protected lysine. This modified approach simplified the purification steps, reduced the reaction time, and produced nanoglobules with a high yield compared to the previously reported synthetic procedure.^{31,32} Immediately after conjugation of the spacer 4-formylbenzoic group to the G₂ nanoglobule, the remaining amino groups were blocked by acetylation to neutralize the G₂ positive charges. We also tested the effect of differing degrees of drug

conjugation on the solubility of the resulting drug conjugates in PBS. The solubility of the nanoglobular 5-ASA conjugate AGFB-ASA decreased significantly when the stoichiometry reached eight drug molecules per nanoglobule. We found that the conjugate with approximately five 5-ASAs per nanoglobule had good aqueous solubility and carried a reasonable drug load.

It was observed that laser energy of MALDI-TOF mass spectrometry had a significant effect on the analysis of the nanoglobular drug conjugate. With relatively high laser energy, the measured mass value (m/z = 5866.58) was smaller than the calculated value based on the drug content determined by proton NMR and UV spectrometry. But with relatively low laser energy, the MALDI-TOF mass spectrum of the conjugate showed a mass close to the calculated mass (Figure 2b). Since Schiff bases are not very stable and acid was used as a matrix, the laser with high energy could have broken the imine bond, resulting in a mass lower than the calculated mass for the conjugate. However, with proper laser energy used for MALDI-TOF mass spectroscopy, similar drug loading results were obtained with three different analytical methods, 5.6 drug molecules per nanoglobule on average from

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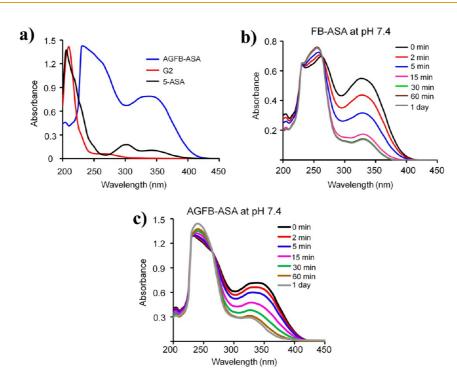


Figure 3. Dynamic UV spectra showing drug release kinetics from the Schiff base conjugate. (a) UV spectra for AGFB-ASA, G₂, and 5-ASA in PBS at pH 7.4; (b) FB-ASA in PBS at pH 7.4; (c) AGFB-ASA in PBS at pH 7.4.

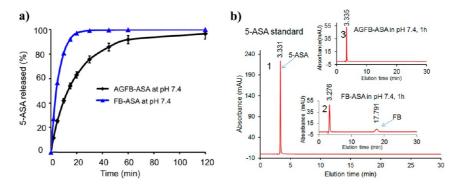


Figure 4. *In vitro* drug release kinetics for the conjugates. (a) Release kinetic profiles of 5-ASA from AGFB-ASA in PBS at pH 7.4 and from FB-ASA in PBS at pH 7.4 assayed by UV spectroscopy; (b) HPLC analyses of the released products: (1) 5-ASA standard, (2) FB-ASA in PBS at pH 7.4 at 1 h, (3) AGFB-ASA in PBS at pH 7.4 PBS. HPLC conditions: analytical C18 reverse column (250 mm \times 4.6 mm, i.d., 5 μ m particle size) with a mobile phase of H₂O/MeOH (80:20, v/v) with 0.05% trifluoroacetic acid, flow rate of 1.0 mL/min, and UV detector set at 303 nm.

¹H NMR, 5.8 from UV spectroscopy, and 5 from mass spectrometry. The size of nanoglobule G_2 determined by dynamic light scattering is about 5 nm as reported in a previous publication.³⁵

Compared to the free Schiff base FB-ASA, AGFB-ASA also had a controllable release profile. This probably was due to steric hindrance around the Schiff base in the nanoglobular conjugate. Although drug release from AGFB-ASA was more controllable than that from the free Schiff base, the overall release rate from AGFB-ASA was still relatively high in aqueous solution. The Schiff base linkage structure could be modified to decrease the hydrolysis rate further and achieve an even more controllable drug release. One possible approach could be to use a ketone group to form a Schiff base with the drug; a steric effect could decrease the hydrolysis rate of this linkage. Another idea would be to introduce an electron-withdrawing group on the benzene ring of 4-formylbenzoic acid to stabilize the Schiff base and decrease the hydrolysis rate. A more controllable and sustained drug release profile would maintain stable drug concentrations *in vivo* and confer more prolonged protective effects on the retina.

To effectively demonstrate the prolonged protective effect of the nanoglobular drug conjugate, we modified the treatment protocol in this study. The *in vivo* circulation half-life of free 5-ASA was noted to be about 2.4 h.⁶ In the previous study, ⁵ free 5-ASA was administrated by oral gavage (2 mg/mouse) 2 h before light exposure, conferring good protection against light-induced retinal degeneration in *Abca4^{-/-}Rdh8^{-/-}* mice. Here, free 5-ASA or the AGFB-ASA conjugate

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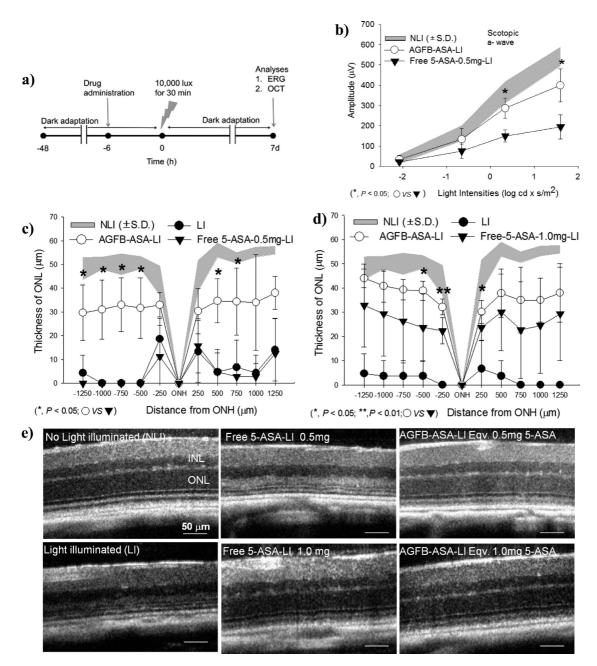


Figure 5. Protective effects of AGFB-ASA and free 5-ASA pretreatment on the development of acute light-induced retinal degeneration in $Abca4^{-/-}Rdh8^{-/-}$ mice (NLI = no light illumination; LI = light illuminated). (a) Schematic representation of the experimental design. After 4-week-old $Abca4^{-/-}Rdh8^{-/-}$ mice were kept in the dark for 48 h, either free 5-ASA or conjugate AGFB-ASA was intraperitoneally injected at a dose equivalent to 0.5 mg or 1.0 mg of 5-ASA per mouse 6 h before light exposure at 10 000 lx for 30 min. Mice were then kept in the dark for 7 days, after which final retinal evaluations were performed. (b) ERG responses of pretreated $Abca4^{-/-}Rdh8^{-/-}$ mice at 4 weeks of age with free 5-ASA or conjugate AGFB-ASA at a dose of 0.5 mg of 5-ASA per mouse prior to light illumination and evaluated 7 days later. ERG responses were recorded under scotopic conditions. Error bars indicate SD of the means (n = 3). ONL thicknesses were measured from OCT images obtained along the horizontal meridian from the nasal to temporal retina: (c) 0.5 mg of 5-ASA per mouse (d) 1.0 mg of 5-ASA per mouse. Statistical analyses were performed by the two-way analysis of variance (Student's *t* test). Error bars indicate SD of the means (n = 4-5). (e) OCT images indicate representative morphology of $Abca4^{-/-}Rdh8^{-/-}$ mouse retinas. Scale bar indicates 50 μ m in the OCT image.

was intraperitoneally injected at a dose equivalent to 0.5 mg or 1.0 mg of 5-ASA per mouse 6 h before light exposure to evaluate the 7-day protective effect of the drug conjugate. Our results showed that the nano-globular drug conjugate had a greater 7-day effect than free 5-ASA at a reduced dose (0.5 mg/per mouse)

in protecting the retinas of 4-week-old $Abca4^{-/}$ $^{-}Rdh8^{-/-}$ mice against bright-light-induced retinal degeneration. This observation indicates that the nano-globular drug conjugate exhibited more prolonged protection against retinal degeneration at a reduced dose due to its controlled and sustained release of

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5-ASA. In contrast, the free drug at the low dose was probably cleared from the body much more rapidly, resulting in a diminished retinal protective effect 7 days after bright light exposure. A more prolonged protective effect could be achieved by chemical modification of nanoglobular conjugates to achieve a more controllable and sustained drug release profile. Currently, we are focusing on structural optimization of nanoglobular drug conjugates to achieve this goal.

CONCLUSION

A water-soluble G_2 nanoglobular conjugate, AGFB-ASA, containing a hydrolyzable Schiff base linkage was

designed to treat retinal degeneration. AGFB-ASA showed a prolonged release of 5-ASA. Upon pretreatment by intraperitoneal injection, the nanoglobular conjugate of 5-ASA produced a greater and more prolonged protective effect at a reduced dose against bright-light-induced retinal degeneration in $Abca4^{-/}$ – $Rdh8^{-/-}$ mice than free 5-ASA. The structure of nanoglobular drug conjugates can be further modified to optimize the release rate of a drug to achieve the best possible therapeutic efficacy. Nanosized nanoglobular 5-ASA conjugates with a degradable spacer show promise in protecting mouse retinas against light-induced retinal degeneration.

MATERIALS AND METHODS

Materials. Octa(3-aminopropyl)silsesquioxane hydrochloride (Octa Ammonium POSS-HCI) came from Hybrid Plastics (Hattiesburg, MS, USA). (Boc)₂-L-lysine-OH and dioxane were purchased from Fisher Scientific (Pittsburgh, PA, USA). Pentafluorophenol (pfp-OH) was obtained from Oakwood Products, Inc. (Columbia, SC, USA). 4-Formylbenzoic acid (FB) and diisopropyl ether were from Sigma-Aldrich (Louis, MO, USA). Anhydrous *N*,*N*-diisopropylethylamine (DIPEA), dichloromethane (DCM), and *N*,*N*-dimethylformamide (DMF) were purchased from Alfa Aesar (Ward Hill, MA, USA). Trifluoroacetic acid (TFA) was obtained from ACROS Organics (Morris Plains, NJ, USA). All reagents were used without further purification.

Synthesis of (Boc)₂-1-lysine-pfp. (Boc)₂-L-lysine-OH (11.3 g, 32.6 mmol), pfp-OH (6.00 g, 32.6 mmol), and *N*,*N'*-dicyclohexyl-carbodiimide (DCC, 6.7 g, 33 mmol) were dissolved in 200 mL of dioxane, and the reaction mixture was stirred at room temperature overnight. After dioxane was removed under vacuum evaporation, the crude product was purified by crystallization twice in disopropyl ether. The yield of (Boc)₂-L-lysine-pfp was 11.0 g, 66.0%. MALDI-TOF (*m*/*z*, [M + H]⁺): 512.47 (calculated), 512.10 (observed).

Synthesis of (I-Lysine)₈-OAS (G₁). Octa(3-aminopropyl)silsesquioxane hydrochloride (OSA · 8HCl, 1.2 g, 1.0 mmol), (Boc)₂-L-lysine-pfp (12.3 g, 24 mmol, 3 equivalents of amino groups), and DIPEA (2.1 mL, 12 mmol, 1.5 equivalents of free amino groups) were dissolved in 25 mL of DMF. The reaction mixture was stirred at room temperature for 24 h under N₂. The product then was precipitated by addition of ice-cold anhydrous diethyl ether. The dried precipitate of [(*t*-BOC)₂-L-lysine]₈-OAS was dissolved in 30 mL of TFA/DCM (1:1) and stirred at room temperature for 4 h to remove t-BOC groups. The resulting solution was concentrated under vacuum to a viscous oil. The residue was treated with ice-cold anhydrous diethyl ether to provide a white solid product. The yield of (L-lysine)₈-OAS trifluoroacetate was 2.94 g (78.8%). MALDI-TOF (*m*/*z*, [M + H]⁺): 1907.20 (calculated for C₇₂H₁₆₀N₂₄O₂₀Si₈), 1907.04 (observed).

Synthesis of (L-Lysine)₁₆-(L-Jysine)₈-OAS (G₂). (L-Lysine)₈-OAS (G₁) trifluoroacetate (2.5 g, 0.67 mmol), (Boc)₂-L-Lysine-pfp (16.5 g, 32.2 mmol), and DIPEA (2.8 mL, 16.1 mmol) were dissolved in 35 mL of DMF. The reaction then was carried out as described for the synthesis of (L-Lysine)₈-OAS. White (L-Lysine)₁₆-(L-Lysine)₈-OAS (G₂) trifluoroacetate was obtained in a yield of 4.95 g, 98.0%. MALDI-TOF (m/z, [M+ H]⁺): 3956.63 (calculated for C₁₆₈H₃₅₂N₅₆O₃₆Si₈); 3956.84 (observed).

Synthesis of G₂-FB. G₂ nanoglobule, $(L-lysine)_{16}-(L-lysine)_8-$ octa(3-aminopropyl)silsesquioxane (OAS) trifluoroacetate (2.0 g, 0.26 mmol), and DIPEA (2.2 mL, 13 mmol) were dissolved in 30 mL of DMF. 4-Formylbenzoic pentafluorophenol-activated ester solution (1.32 mmol, 5 equivalents to G₂) in 60 mL of DMF was slowly dripped into the G₂ solution, and the mixture was allowed to react in an ice-bath for 6 h. The filtered solution then was dripped into ice-cold ethyl ether, after which precipitates

were collected and dried under vacuum to give compound **1** (G₂-FB). The yield of G₂-FB was 1.12 g, 92.1%. MALDI-TOF (*m/z*, $[M + Na]^+$): 4639.24 (observed), 4640.21 (calculated from G₂ with 5 formylbenzoic acid groups).

Acetylation of G₂-FB to Block Free Amine Groups on G₂ (AGFB). G₂-FB (0.724 g, 0.156 mmol) was dissolved in 60 mL of distilled water. Acetic anhydride (4.7 mL, 10 times the equivalent of free amino groups) was dripped in, and the pH was maintained at 8.0 with triethylamine. The reaction was kept in an ice-bath for 4 h. Then the reaction solution was dialyzed with a membrane with a molecular weight cutoff of 0.5–1.0 kDa overnight and lyophilized to obtain compound AGFB. The yield of AGFB was 0.52 g, 59.9%. MALDI-TOF (m/z, [M + Na]⁺): 5773.71 (observed), 5774.29 (calculated from acetylated G₂ with 5 formylbenzoic acid groups).

Synthesis of AGFB-ASA. AGFB (0.51 g, 0.088 mmol) was dissolved in 10 mL of methanol. 5-ASA (0.15 g, 0.98 mmol, 2 times the equivalent of aldehyde groups) and a drop of acetic acid were added. The reaction was kept at room temperature overnight. Then the filtered solution was dropped into 100 mL of cold ethyl ether, and precipitates were collected by centrifugation and redissolved in methanol. The crude product was purified by running it through a Sephadex G-15 column to remove free 5-ASA, and the first yellow-brownish eluate was collected and dried under vacuum to give the final product compound AGFB-ASA. The yield of AGFB-ASA was 0.24 g, 42.4%. MALDI-TOF $(m/z, [M + Na]^+)$: 6020.91 (observed), 6428.80 (calculated), from acetylated G₂ conjugated with five drug molecules (C₂₉₇H₄₅₁N₆₁O₈₃Si₈). ¹H NMR (300 MHz, CD₃OD, ppm): 8.63 (br, 5H, -CHN-), 8.07 and 7.93 (dm, 35H, -CH on benzene ring), 4.40-4.21 (dm, 24H, -CH), 3.50-2.80 (br, 16H, -(CH₂)₃CH₂; 16H, -Si(CH₂)₂CH₂; 32H, -(CH₂)₃CH₂NH₂), 1.76 (d, 81H, -CH₃O), 1.70-1.25 (br m, 32H, -CHCH₂; 16H, -CHCH₂; 32H, -(CH₂)₂CH₂CH₂; 16H, -(CH₂)₂CH₂CH₂; 32H, -CH₂CH₂(CH₂)₂, 16H, -CH₂CH₂(CH₂)₂; 16H, -SiCH₂CH₂), 0.63 (br, 16H, -SiCH₂).

Synthesis of FB-ASA. 4-Formylbenzoic acid (0.45 g, 3.0 mmol) and 5-ASA (0.15 g, 1.0 mmol) were added to 15 mL of methanol. Acetic acid (4 μ L) was then added to the reaction mixture, which turned yellow after 10 min and was stirred at room temperature for 24 h. Precipitates were collected by centrifugation, washed with methanol three times, and dried under vacuum to provide the final product FB-ASA. The yield of FB-ASA was 240 mg, 84.2%. MALDI-TOF (m/z, [M + H]⁺): 285.25 (calculated for C₁₅H₁₁NO₅), 285.60 (observed). ¹H NMR (400 MHz, (CD₃)₂SO, ppm): 8.78 (s, 1H, -CHN-), 8.03 (d, 4H, -CH), 7.76 (d, 1H, -CH), 7.61 (dd, 1H, -CH), 7.03 (d, 1H, -CH).

5-ASA Release Experiments. Release of 5-ASA from the G₂ conjugate by hydrolysis of the Schiff base was evaluated by using UV spectroscopy. Release of 5-ASA from 0.01 mM AGFB-ASA was measured in 0.05 M phosphate buffers at pH 7.4. Release of 5-ASA from 0.05 mM FB-ASA in pH 7.4 PBS was also tested as a control. Dendrimer conjugate solution or FB-ASA solution was incubated at 37 °C, and samples were taken at selected time intervals up to 24 h. Immediately after sampling at

each time point, samples were scanned from 200 to 450 nm with a SpectraMax M5 spectrometer (Molecular Devices, Sunnyvale, CA, USA). Released drug from conjugates AGFB-ASA and FB-ASA in 0.05 M phosphate buffer, pH 7.4, after 1 h was analyzed by HPLC (HPLC conditions: analytical C18 reverse column (250 mm \times 4.6 mm, i.d., 5 μ m particle size) eluted with a mobile phase of H2O/MeOH (80:20, v/v) with 0.05% trifluoroacetic acid, flow rate of 1.0 mL/min, and UV detector set at 303 nm).

Animal Studies. With 4 or 5 animals used for each treatment group, 4-week-old $Abca4^{-/-}Rdh8^{-/-}$ mice⁴ were kept in the dark for 48 h before each experiment. Then free 5-ASA (0.5 and 1.0 mg per mouse) or conjugate AGFB-ASA (4.3 and 8.6 mg per mouse, equivalent to 0.5 and 1.0 mg of free 5-ASA per mouse) was intraperitoneally injected 6 h before light exposure at 10 000 lx for 30 min. Mice then were kept in the dark for 7 days, after which final retinal evaluations were performed. Mice were anesthetized by intraperitoneal injection of a cocktail (20 uL g body weight) containing ketamine (6 mg mL⁻¹) and xylazine (0.44 mg mL⁻¹) in 10 mM sodium phosphate, pH 7.2, and 100 mM NaCl. Pupils were dilated with 0.01% tropicamide. Electroretinograms were recorded as previously reported.² All experimental procedures were performed under a safety light. A contact lens electrode was placed on the eye, and a reference electrode and ground electrode were placed underneath the skin between the two ears and in the tail, respectively. ERGs were recorded with the universal electrophysiologic system UTAS E-3000 (LKC Technologies, Inc., Gaithersburg, MD, USA). The light intensity calibrated by the manufacturer was computer-controlled. Mice were placed in a Ganzfeld dome, and scotopic responses to flash stimuli were obtained from both eyes simultaneously. Retinas of mice were then imaged in vivo with ultra-high-resolution spectral-domain OCT (SD-OCT; Bioptigen, Irvine, CA) 24 h after the ERG test. Dark-adapted mice were anesthetized according to the same protocol used for ERG. Five pictures acquired in the B-scan mode were used to construct each final averaged SD-OCT image. Quantitative morphometric obtained from ONL thickness was measured from OCT images along the horizontal meridian from the nasal to temporal retina.

Conflict of Interest: The authors declare no competing financial interest.

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REFERENCES AND NOTES

- Maeda, T.; Golczak, M.; Maeda, A. Retinal Photodamage Mediated by All-Trans-Retinal. Photochem. Photobiol. 2012, 88, 1309–1319.
- Kiser, P. D.; Golczak, M.; Maeda, A.; Palczewski, K. Key Enzymes of the Retinoid (Visual) Cycle in Vertebrate Retina. *Biochim. Biophys. Acta* 2012, *1821*, 137–151.
- 3. Palczewski, K. Retinoids for Treatment of Retinal Diseases. Trends Pharmacol. Sci. 2010, 31, 284–295.
- 4. Maeda, A.; Maeda, T.; Golczak, M.; Palczewski, K. Retinopathy in Mice Induced by Disrupted All-*Trans*-Retinal Clearance. *J. Biol. Chem.* **2008**, *283*, 26684–26693.
- Maeda, A.; Golczak, M.; Chen, Y.; Okano, K.; Kohno, H.; Shiose, S.; Ishikawa, K.; Harte, W.; Palczewska, G.; Maeda, T.; et al. Primary Amines Protect against Retinal Degeneration in Mouse Models of Retinopathies. *Nat. Chem. Biol.* 2012, *8*, 170–178.
- Klotz, U.; Maier, K. E. Pharmacology and Pharmacokinetics of 5-Aminosalicylic Acid. *Dig. Dis. Sci.* 1987, 32, S46–S50.
- 7. Gaudana, R.; Ananthula, H. K.; Parenky, A.; Mitra, A. K. Ocular Drug Delivery. *AAPS J.* **2010**, *12*, 348–360.
- Geroski, D. H.; Edelhauser, H. F. Drug Delivery for Posterior Segment Eye Disease. *Invest. Ophthalmol. Vis. Sci.* 2000, 41, 961–964.
- Del Amo, E. M.; Urtti, A. Current and Future Ophthalmic Drug Delivery Systems. A Shift to the Posterior Segment. Drug Discovery Today 2008, 13, 135–143.

- 10. Eljarrat-Binstock, E.; Pe'er, J.; Domb, A. J. New Techniques for Drug Delivery to the Posterior Eye Segment. *Pharm. Res.* **2010**, *27*, 530–543.
- Chastain, J. E. General Considerations in Ocular Drug Delivery. In *Ophthalmic Drug Delivery Systems*; Mitra, A. K., Ed.; Marcel Dekker, Inc.: New York, 2003; pp 59–107.
- Edelhauser, H. F.; Rowe-Rendleman, C. L.; Robinson, M. R.; Dawson, D. G.; Chader, G. J.; Grossniklaus, H. E.; Rittenhouse, K. D.; Wilson, C. G.; Weber, D. A.; Kuppermann, B. D.; *et al.* Ophthalmic Drug Delivery Systems for the Treatment of Retinal Diseases: Basic Research to Clinical Applications. *Invest. Ophthalmol. Vis. Sci.* **2010**, *51*, 5403–5420.
- Babu, V. R.; Mallikarjun, V.; Nikhat, S. R.; Srikanth, G. Dendrimers: A New Carrier System for Drug Delivery. *Int. J. Pharm. Appl. Sci.* **2010**, *1*, 1–10.
- Vandamme, T. F. Microemulsions as Ocular Drug Delivery Systems: Recent Developments and Future Challenges. Prog. Retin. Eye Res. 2002, 21, 15–34.
- Kaur, I. P.; Garg, A.; Singla, A. K.; Aggarwal, D. Vesicular Systems in Ocular Drug Delivery: An Overview. *Int. J. Pharm.* 2004, 269, 1–14.
- Barbu, E.; Verestiuc, L.; Iancu, M.; Jatariu, A.; Lungu, A.; Tsibouklis, J. Hybrid Polymeric Hydrogels for Ocular Drug Delivery: Nanoparticulate Systems from Copolymers of Acrylic Acid-Functionalized Chitosan and N-Isopropylacrylamide or 2-Hydroxyethyl Methacrylate. *Nanotechnology* 2009, 20, 225108.
- Hacker, M. C.; Haesslein, A.; Ueda, H.; Foster, W. J.; Garcia, C. A.; Ammon, D. M.; Borazjani, R. N.; Kunzler, J. F.; Salamone, J. C.; Mikos, A. G. Biodegradable Fumarate-Based Drug-Delivery Systems for Ophthalmic Applications. J. Biomed. Mater. Res. A 2009, 88, 976–989.
- Jiang, J.; Moore, J. S.; Edelhauser, H. F.; Prausnitz, M. R. Intrascleral Drug Delivery to the Eye Using Hollow Microneedles. *Pharm. Res.* 2009, *26*, 395–403.
- 19. Kim, Y. C.; Park, J. H.; Prausnitz, M. R. Microneedles for Drug and Vaccine Delivery. *Adv. Drug Delivery Rev.* **2012**, *64*, 1547–1568.
- 20. Shirasaki, Y. Molecular Design for Enhancement of Ocular Penetration. *J. Pharm. Sci.* **2008**, *97*, 2462–2496.
- 21. Wadhwa, S.; Paliwal, R.; Paliwal, S. R.; Vyas, S. P. Nanocarriers in Ocular Drug Delivery: An Update Review. *Curr. Pharm. Des.* **2009**, *15*, 2724–2750.
- 22. Kuno, N.; Fujii, S. Recent Advances in Ocular Drug Delivery Systems. *Polymers* **2011**, *3*, 193–221.
- Cheng, Y. Y.; Xu, Z. H.; Ma, M. L.; Xu, T. W. Dendrimers as Drug Carriers: Applications in Different Routes of Drug Administration. J. Pharm. Sci. 2008, 97, 123–143.
- Medina, S. H.; El-Sayed, M. E. H. Dendrimers as Carriers for Delivery of Chemotherapeutic Agents. *Chem. Rev.* 2009, 109, 3141–3157.
- Durairaj, C.; Kadam, R. S.; Chandler, J. W.; Hutcherson, S. L.; Kompella, U. B. Nanosized Dendritic Polyguanidilyated Translocators for Enhanced Solubility, Permeability, and Delivery of Gatifloxacin. *Invest. Ophthalmol. Vis. Sci.* 2010, *51*, 5804–5816.
- Vandamme, T. F.; Brobeck, L. Poly(amidoamine) Dendrimers as Ophthalmic Vehicles for Ocular Delivery of Pilocarpine Nitrate and Tropicamide. *J. Controlled Release* 2005, *102*, 23–38.
- Kambhampati, S. P.; Kannan, R. M. Dendrimer Nanoparticles for Ocular Drug Delivery. J. Ocul. Pharmacol. Ther. 2013, 29, 151–165.
- Yang, H.; Tyagi, P.; Kadam, R. S.; Holden, C. A.; Kompella, U. B. Hybrid Dendrimer Hydrogel/PLGA Nanoparticle Platform Sustains Drug Delivery for One Week and Antiglaucoma Effects for Four Days Following One-Time Topical Administration. ACS Nano 2012, 6, 7595–7606.
- lezzi, R.; Guru, B. R.; Glybina, I. V.; Mishra, M. K.; Kennedy, A.; Kannan, R. M. Dendrimer-Based Targeted Intravitreal Therapy for Sustained Attenuation of Neuroinflammation in Retinal Degeneration. *Biomaterials* 2012, 33, 979–988.
- Holden, C. A.; Tyagi, P.; Thakur, A.; Kadam, R.; Jadhav, G.; Kompella, U. B.; Yang, H. Polyamidoamine Dendrimer Hydrogel for Enhanced Delivery of Antiglaucoma Drugs. *Nanomedicine* **2012**, *8*, 776–783.

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- 31. Kaneshiro, T. L.; Wang, X.; Lu, Z. R. Synthesis, Characterization, and Gene Delivery of Poly-L-lysine Octa(3-aminopropyl)silsesquioxane Dendrimers: Nanoglobular Drug Carriers with Precisely Defined Molecular Architectures. Mol. Pharmaceutics 2007, 4, 759-768.
- 32. Kaneshiro, T. L.; Jeong, E. K.; Morrell, G.; Parker, D. L.; Lu, Z. R. Synthesis and Evaluation of Globular Gd-DOTA-Monoamide Conjugates with Precisely Controlled Nanosizes for Magnetic Resonance Angiography. Biomacromolecules 2008, 9, 2742-2748.
- 33. Kratz, F.; Beyer, U.; Schutte, M. T. Drug-Polymer Conjugates Containing Acid-Cleavable Bonds. Crit. Rev. Ther. Drug Carrier Syst. 1999, 16, 245-288.
- 34. Saito, H.; Hoffman, A. S.; Ogawa, H. I. Delivery of Doxorubicin from Biodegradable PEG Hydrogels Having Schiff Base Linkages. J. Bioact. Compat. Polym. 2007, 22, 589-601.
- 35. Tan, M. Q.; Wu, X. M.; Jeong, E.-K.; Chen, Q. J.; Lu, Z. R. Peptide-Targeted Nanoglobular Gd-DOTA Monoamide Conjugates for Magnetic Resonance Cancer Molecular Imaging. Biomacromolecules 2010, 11, 754–761.
- 36. Maeda, A.; Maeda, T.; Golczak, M.; Imanishi, Y.; Leahy, P.; Kubota, R.; Palczewski, K. Effects of Potent Inhibitors of the Retinoid Cycle on Visual Function and Photoreceptor Protection from Light Damage in Mice. Mol. Pharmacol. 2006, 70, 1220-1229.



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